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Umabanol, a New Tetracyclic Diterpene from a Marine Sponge

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(Received February 10, 1997; CL-970099)

A new diterpene, umabanol, has been isolated from the sponge *Epipolasis kushimotoensis* and its structure determined by spectroscopic analysis. Umabanol possesses a new tetracyclic ring system related to verrucosane class diterpenes.

Verrucosane and related diterpenes have been known as characteristic metabolites of some species of terrestrial plants.¹ For example, Neoverrucosanol (1) has been reported from the liverwort *Mylia verrucosa*² and homoverrucosanol (2) from *Schistochila acuminata*.^{3,4} Recently these two compounds have also been described as the constituents of a marine animal, the sponge *Axinyssa aplysinoides*.⁵ It was the first report on the occurrence of verrucosane-class diterpenes in a marine organism.¹³ In our search for bioactive compounds from marine organisms, we recently examined the constituents of the sponge *Epipolasis kushimotoensis*⁶ and found, in addition to four known verrucosanes (1-4), a new related diterpene named umabanol (5) which was composed of a new tetracyclic ring system. In this paper we describe the isolation of these compounds and the structure elucidation of 5.

A sample (430 g) of the sponge *E. kushimotoensis* was extracted by steeping in acetone. An ethyl acetate soluble portion of the extract was separated by vacuum flash chromatography (silica gel, hexane-EtOAc) followed by reverse phase HPLC (RP-8, MeCN) to give compounds 1 (366 mg), 2 (238 mg), 3 (132 mg), 4 (26 mg), and 5 (9 mg).⁷ Compounds 1-4 were identified to be the known neoverrucosanol, homoverrucosanol, 13-*epi*-neoverrucosanol (3),⁸ and 13-*epi*-homoverrucosanol (4),⁹ respectively, by spectroscopic comparison with those reported.

The molecular formula C₂₀H₃₄O of **5**, $[\alpha]p$ -78° (c 0.10, CHCl₃), was determined by HREIMS (m/z 290.2632, Δ +2.4 mmu).¹⁰ Four sites of unsaturation required by the formula and the absence of olefinic signals in both ¹H and ¹³C NMR spectra indicated the compound to have four rings. The presence of a tertiary hydroxyl group was inferred from an IR (CCl₄)

absorption band at 3620 cm⁻¹ and a ¹³C NMR signal at δ 71.1 (s). The gross structure was determined by 2D NMR connectivity study and comparison of the data with those of 1 and 2. As a result, all the signals of 5 could be assigned as shown in Table 1. Nearly identical NMR data for the portion of C9-C17 and C20 with those of 1 and 2 suggested that 5 had the same partial structure including the 5- and 6-membered rings. Observation of COSY cross signals for H1/H14, H1/H2, H2/H3 α , H2/H3 β , H2/H5, H5/H6α, and H5/H6β together with HMBC results (Table 1) indicated that the remaining part of the structure was composed of another 5-membered ring and a 4-membered ring. The presence of the cyclobutane ring was further substantiated by typical coupling constants, i.e., a long range coupling (J = 1 Hz)between H3 α and H5, and two cis couplings (J = 8 Hz) between H2 and $H3\alpha$ and between H2 and H5. The relative stereochemistry was established by difference NOEs observed for H1/H13, H1/H20, H2/H3\alpha, H2/H5, H2/H14, H2/H15, H2/H19, $H3\alpha/H18$, $H5/H6\alpha$, H5/H18, H5/H19, and H14/H19. Biogenetically 5 may be derived from 1, and vice versa. A question that 5 may be an artifact of the isolation is ruled out by the fact that acid treatment¹¹ of 1 gives 2 in quantitative yield, but no trace of 5. This acid-catalyzed rearrangement has previously been reported by Matsuo et al.² The absolute configuration of 1 has been established by X-ray analysis.² Co-occurrence of 5 with 1 suggests that umabanol has the absolute configuration shown in the structure. The tetracyclic ring system of umabanol is new in diterpenes.

Homoverrucosanol (2) showed moderate cytotoxicity (ICso 2.5-5 μ g/mL) against P388, A549, and HT29 tumor cells, while that of neoverrucosanol (1) was weak (ICso >10 μ g/mL). No cytotoxicity test has been performed with compounds 3-5.

This work was partially supported by Grant-in-Aid for Scientific Research from Ministry of Education, Culture, Sports, and Science of Japan to J.T. (No. 04771880 and 05771931). We thank Prof. D. J. Faulkner, University of California at San Diego, and Prof. C.-L. Wu, Tamkang University, Taiwan, for the NMR data of compounds 1 and 2.

References and Notes

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- 6 The sponge was collected at -30 m by hand using SCUBA at Umabana, Yonaguni Island, Okinawa. Taxonomic identification was carried out by Dr. J. N. A. Hooper, Queensland Museum, South Brisbane, Australia, and a voucher specimen (G304924) is deposited at the museum.
- 7 1: mp 138-140 °C (acetone), $[\alpha]_D$ +2.6° (c 2.8, CHCl₃).

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Table 1. ¹H (500 MHz) and ¹³C (125 MHz) NMR Data for Neoverrucosanol (1), Homoverrucosanol (2), and Umabanol (5) in CDCl₃

	1		2		5		
C#	¹³ C	¹H	¹³ C	¹H	¹³ C	H^{I}	HMBC
1	47.6 d	1.02 dd, J = 4, 12 Hz	43.8 d	2.20 brdd, $J = 5$, 12 Hz	55.1 d	1.31 dd, $J = 9$, 12 Hz	C2,3,7,19
2	25.6 d	0.85 ddd, J = 4, 5, 8 Hz	131.7 d	5.29 brd, $J = 5$ Hz	30.9 d	2.03 ddt, J = 3, 9, 8 Hz	C1,5,14
3	19.6 t	0.28 t, J = 5 Hz	131.3 s	-	41.6 t	1.57 m	
		0.56 dd, J = 5, 8 Hz				2.23 ddd, J = 1, 8, 13 Hz	C1,2,5,18
4	22.0 s	-	42.6 t	2.03 m	71.1 s	-	
				2.54 ddd, J = 2, 11, 13 Hz			
5	71.2 d	4.03 brdd, J = 7, 11 Hz	65.6 d	3.60 ddt, J = 2, 3, 11 Hz	48.2 d	2.61 brq, $J = 8 \text{ Hz}$	C1,2,3,6,18
6	47.2 t	0.68 dd, J = 11, 13 Hz	58.9 t	$1.47 \mathrm{dd}, J = 11, 13 \mathrm{Hz}$	34.3 t	1.45 m	
		1.69 dd, $J = 7$, 13 Hz		1.89 ddd, $J = 2$, 3, 13 Hz		1.54 m	
7	37.1 s	-	38.0 s	-	46.7 s	-	
8	35.3 t	1.10 ddd, J = 2, 4, 13 Hz	39.0 t	1.23 dt, $J = 14$, 3 Hz	38.8 t	1.47 m	
		1.29 dt, $J = 4$, 13 Hz		$1.54 \mathrm{dt}, J = 4, 14 \mathrm{Hz}$		1.57 m	
9	34.6 t	1.35 m	35.1 t	1.32 dt, J = 4, 13 Hz	35.1 t	1.33 m	
		1.42 m		1.40 m		1.43 m	
10	44.0 s	-	42.6 s	-	43.5 s	-	
11	39.2 t	1.05 m	38.8 t	1.03 brq, $J = 11 \text{ Hz}$	39.2 t	1.07 brq, $J = 10 \mathrm{Hz}$	C9,10,12,20
		1.37 m		1.37 m		1.44 m	
12	21.6 t	1.40 m,	21.4 t	1.38 m	20.8 t	1.37 m	
		$1.60 \mathrm{dq}, J = 13, 10 \mathrm{Hz}$		$1.60 \mathrm{dq}, J = 13, 11 \mathrm{Hz}$		1.52 m	
13	45.2 d	1.90 dddd, $J = 4$, 7, 10, 12 Hz	46.8 d	1.71 ddt, $J = 3$, 7, 11 Hz	44.2 d	1.74 m	C1,14,16
14	46.9 d	1.21 t, $J = 12 \text{ Hz}$	47.8 d	1.16 t, $J = 11 \text{ Hz}$	47.6 d	1.13 t, J = 12 Hz	C1,7,10,13
15	28.7 d	2.15 m	28.0 d	2.05 m	27.4 d	1.67 m	C12
16		0.82 d, J = 7 Hz		0.83 d, J = 7 Hz		0.75 d, J = 7 Hz	C13,15,17
17		0.91 d, J = 7 Hz	22.9 q	0.86 d, J = 7 Hz	22.8 q	0.85 d, J = 7 Hz	C13,15,16
18	25.8 q	1.19 s	25.8 q	1.76 s	31.3 q	1.36 s	C3,4,5
19	17.2 q	0.83 s	20.1 q	0.86 s	19.3 q	0.66 s	C1,6,7,8
20	18.6 q	0.74 s	18.1 q	0.78 s	18.6 q	0.78 s	C9,10,11,14

Since the sign of the optical rotation of this sample was opposite to that (-10°) reported,² we determined the absolute configuration of our sample by Modified Mosher's method¹² and found that it was identical with that reported by X-ray analysis. 2: mp 135-136 °C (MeOH), $[\alpha]_D$ +21° (c 0.59, CHCl₃). 3: mp 133-135 °C (MeOH), $[\alpha]_D$ +57° (c 0.62, CHCl₃). 4: $[\alpha]_D$ +43°. Since full NMR data for 1 and 2 have not been reported, these data are listed in Table 1 along with those of 5 for comparison. In addition to diterpenes 1-5, five known sesquiterpenes, (1S,6S,7R,10R)-10-isothiocyanato-4-cadinene (356 mg), $(1S^*,4S^*,5R^*,6S^*,7S^*,10S^*)$ -1-isothiocyanatoaromadendrane (50 mg), epipolasin A (14 mg), β-cubebene (11 mg), and $(1S^*,4R^*,5S^*,10S^*)$ -10-isothiocyanatoguaia-6-ene (6 mg), were isolated from other fractions

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- 11 A solution of 1 (20.2 mg) in acetone was treated with 0.5 N H₂SO₄ as described^{2,9} previously to give 19.0 mg of 2 as a single product which was identical with a sample isolated from the sponge.
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